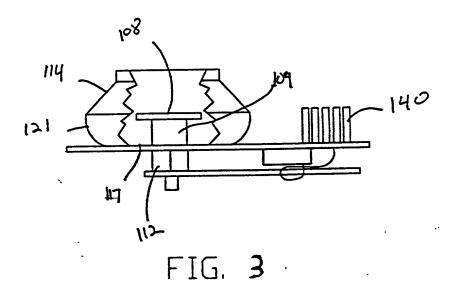
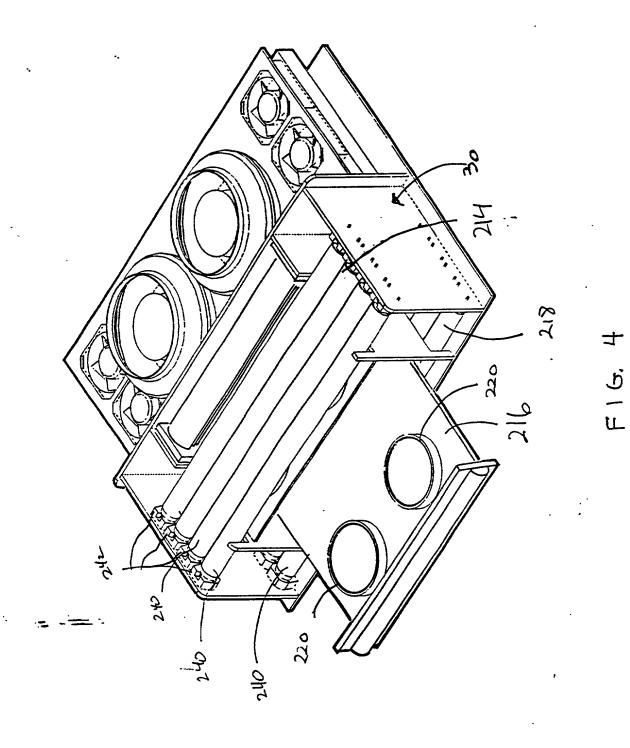
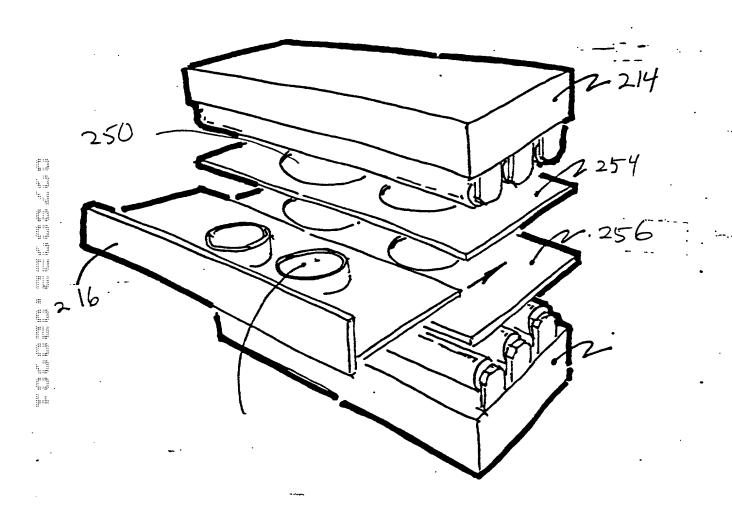


FIG. 2







F1 G. 5

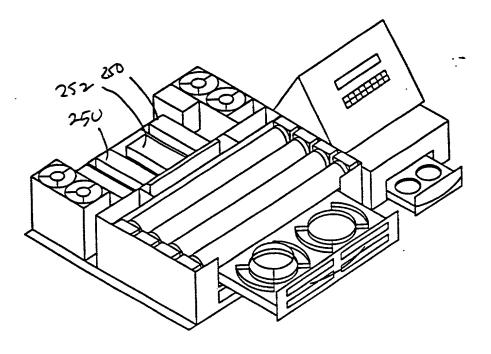
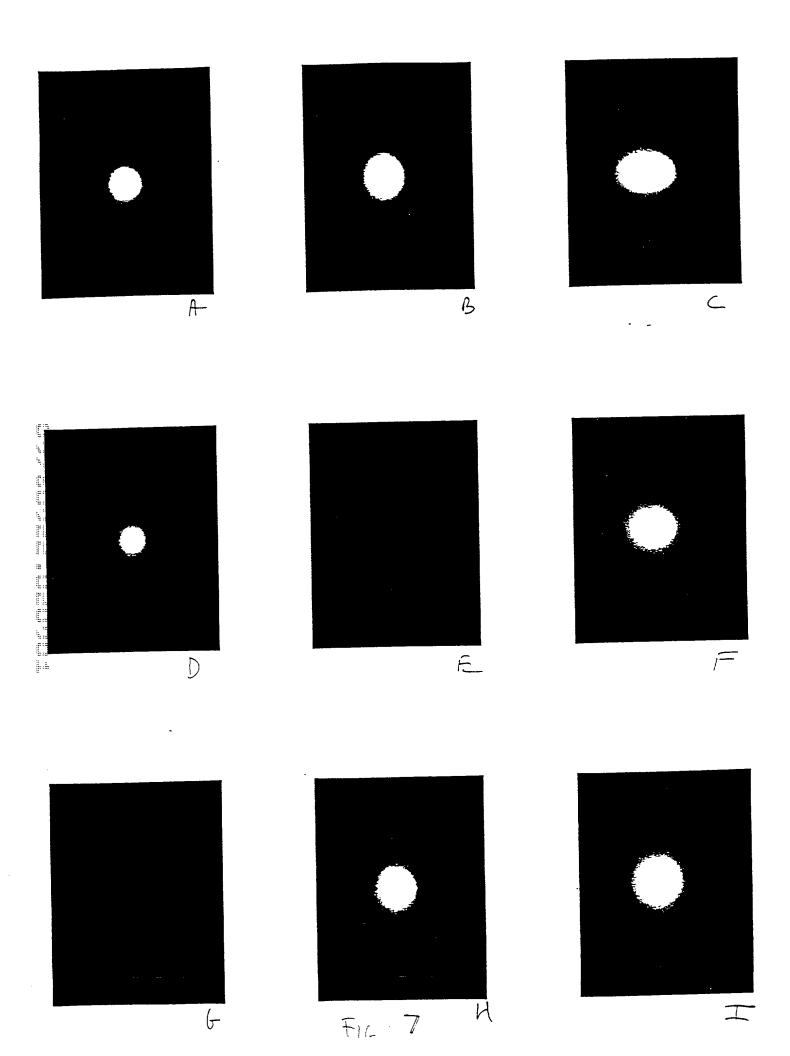
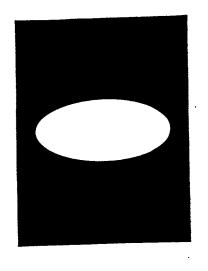
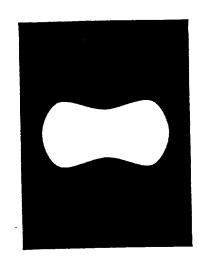
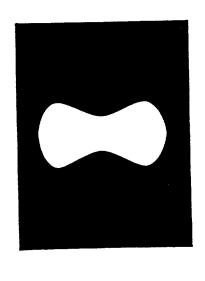


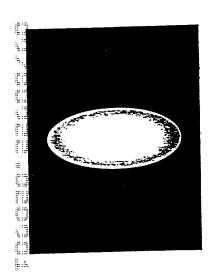
FIG. 6

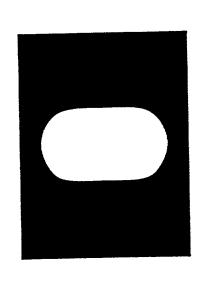


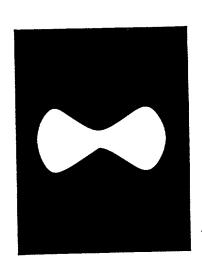


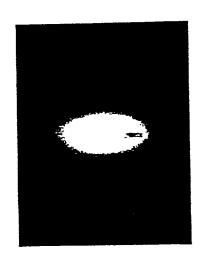


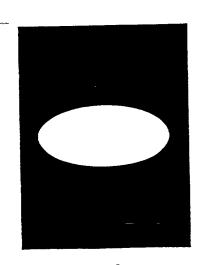


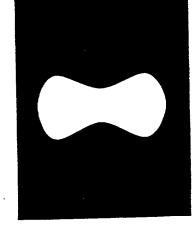




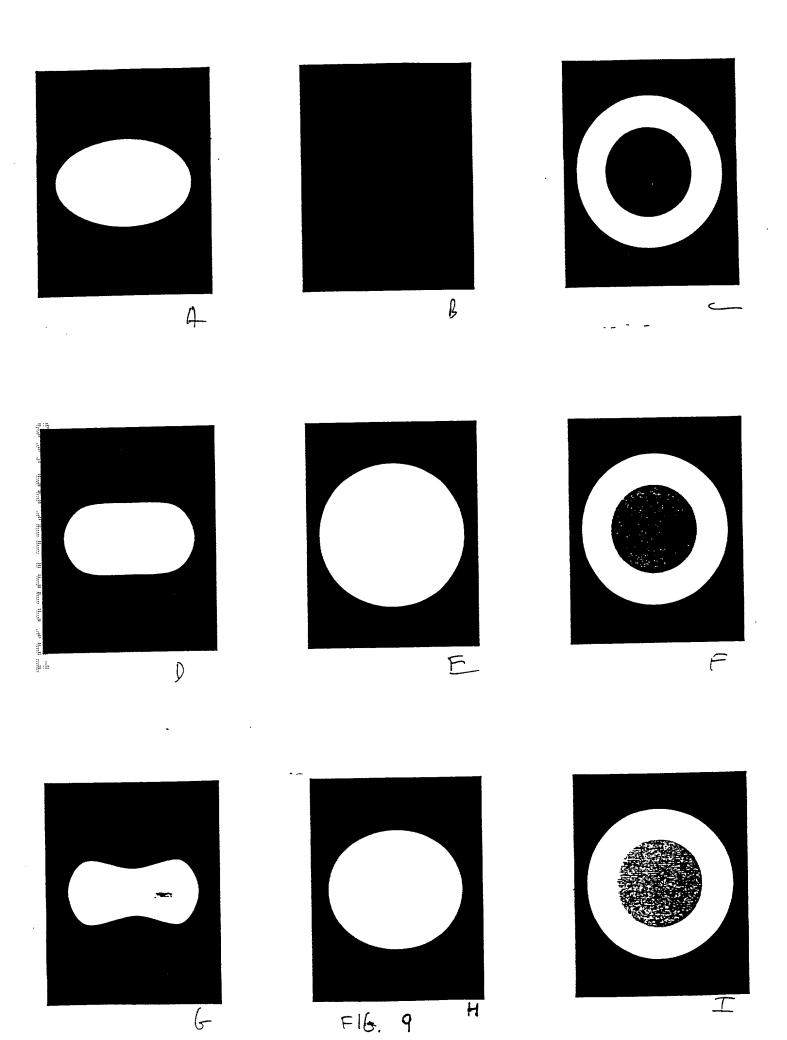


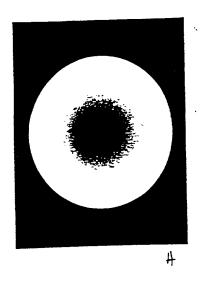


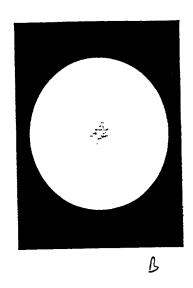


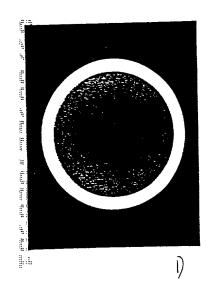


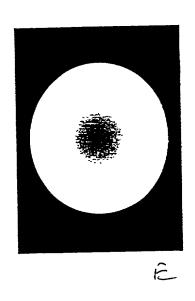
F16 8

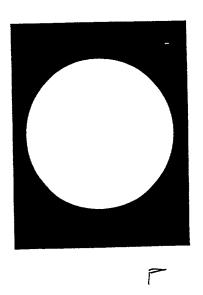


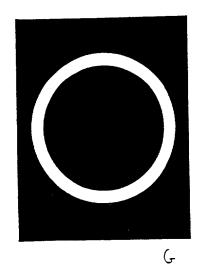


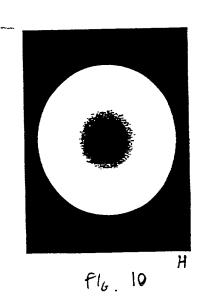


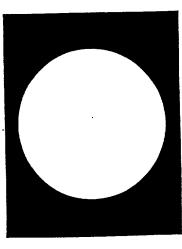




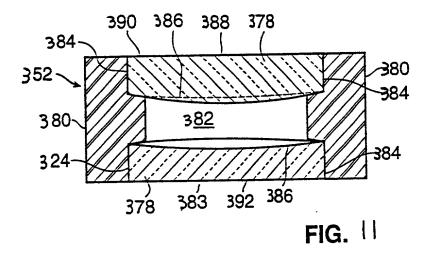


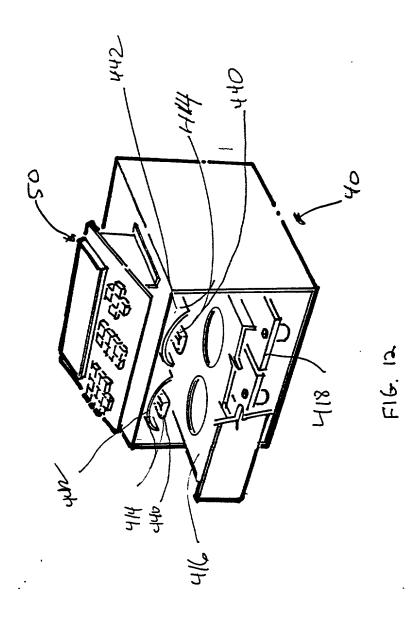






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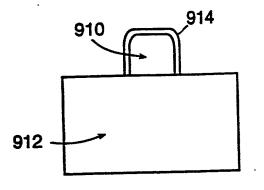


FIG. 13

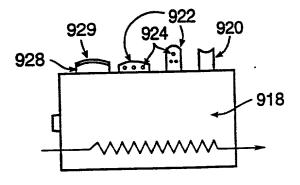


FIG. 14 /

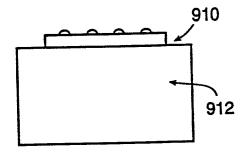


FIG. 15

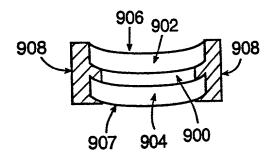
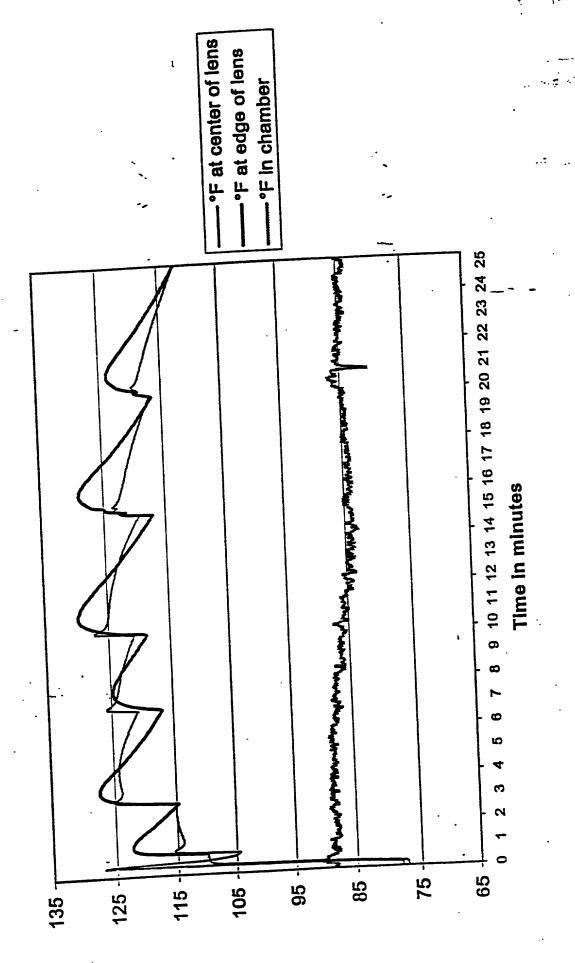


FIG. 16





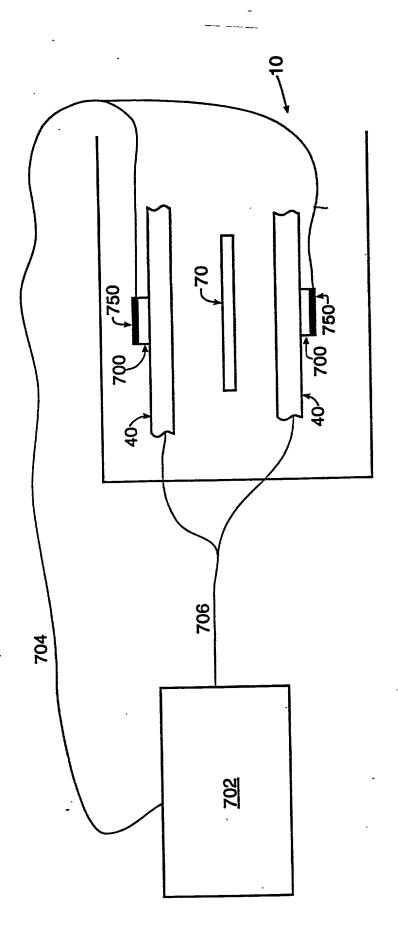
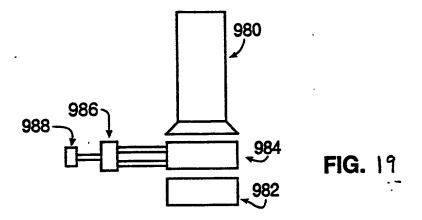


표. _ 8

• •



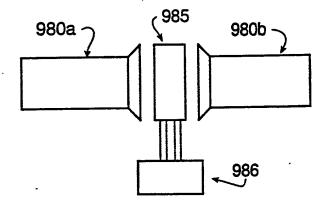


FIG. へ

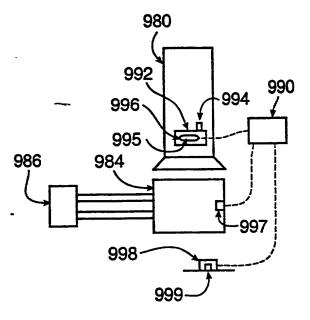
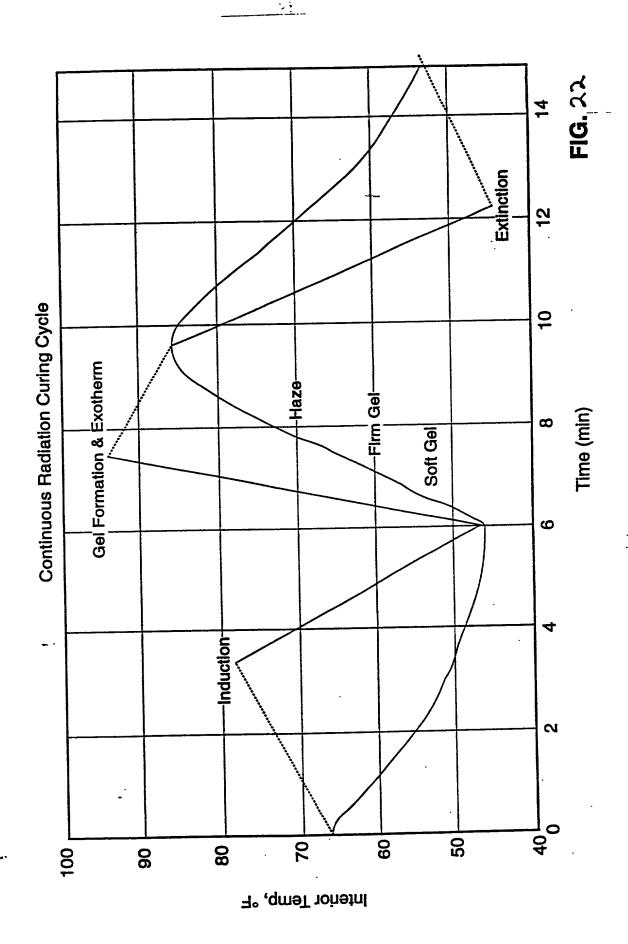


FIG. 기



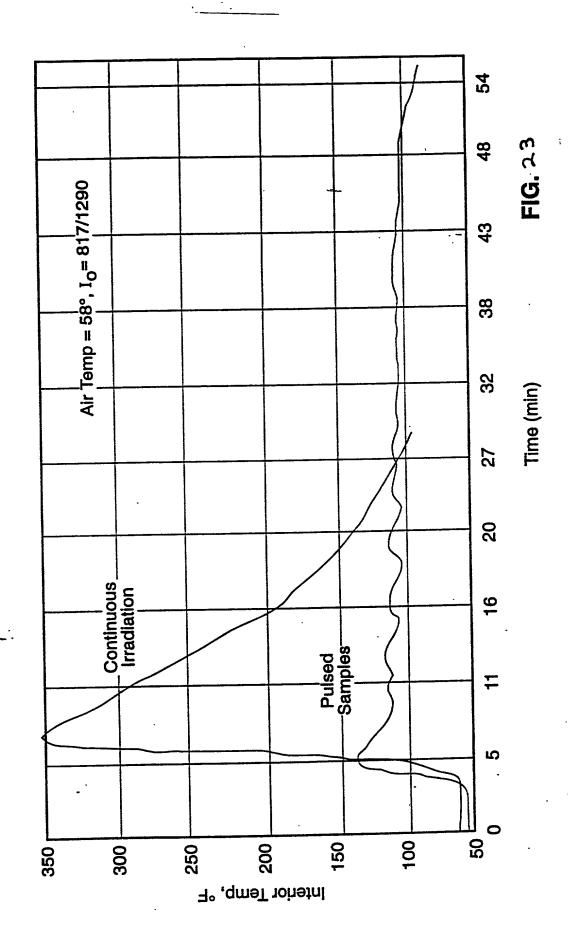


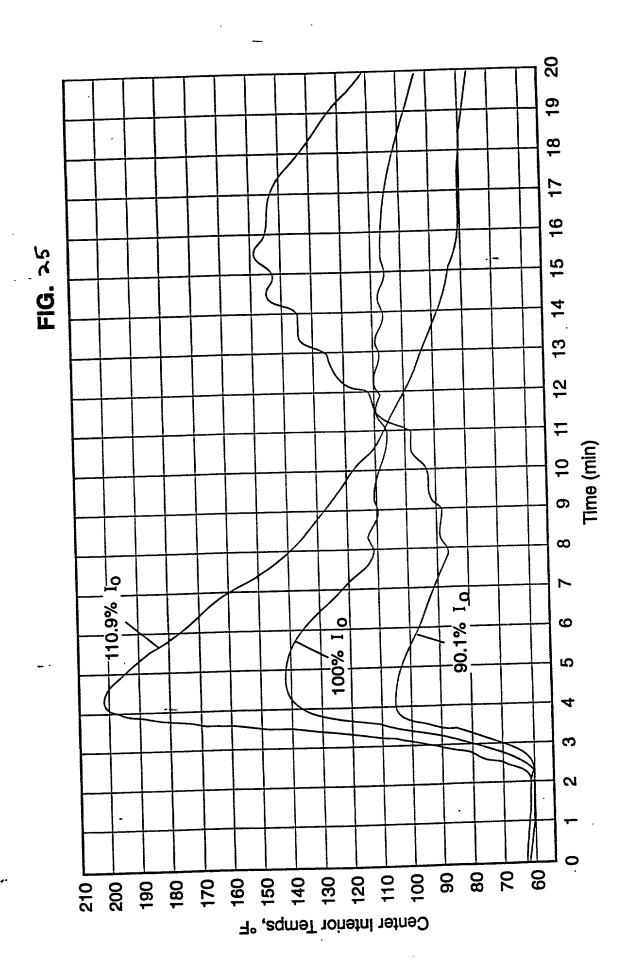
FIG. 2 4 IDENTITY OF MONOMER	Differences in inhibitor & initiator levels between batches of otherwise identical monomers may significantly affect induction periods. Various radiation curable compounds may also vary widely in their preferred initial exposure times due to inherent differences in their reactivity.	A significant effect that various monomers may have upon total cycle time will come from their different preferred initial exposure times.	The duration of the pulses may be adjusted to create the desired amount of reaction and heat generation for the for the particular tens forming material being cured. Adjusting the cooling period between pulses may also be beneficial.
Method Variables RATE OF COOLING		Increased rates of heat removal may allow for a reduction in the time between pulses and thus total cycle time.	Increased rates of heat removal tend to allow for a reduction in the time between pulses.
Interaction of Pulsed Method Variables LIGHT INTENSITY RATE OF COOL	As light intensity increases, initial exposure time may tend to decrease. The light intensity level in may be controlled for a fixed curing cycle and initial exposure time. It is believed, however, that changes in light intensities may have little impact above a certain light "saturation" point for the sample.	ı." to	a given light intensity level, duration of the pulses may adjusted to create the desired ount of reaction. The timing ween the pulses may also be adjusted.
The effect that this variable will tend to have: MASS OF SAMPLE	initial / sased. eracts or nine a lime.	Increased sample mass may require increased light intensity may require increased total cycle time cause a decrease in the initial to dissipate the additional heat exposure period. It is believed however that changes in light point for the sample.	Increased sample mass may require longer periods of cooling the between pulses of light. More be theat tends to be generated from ameach pulse for larger samples, bel thus requiring longer time periods so to remove heat.
The effect that th	On this cycle variable in: OPTIMAL INITIAL EXPOSURE TIME	TOTAL CYCLE TIME	TIMING BETWEEN PULSES

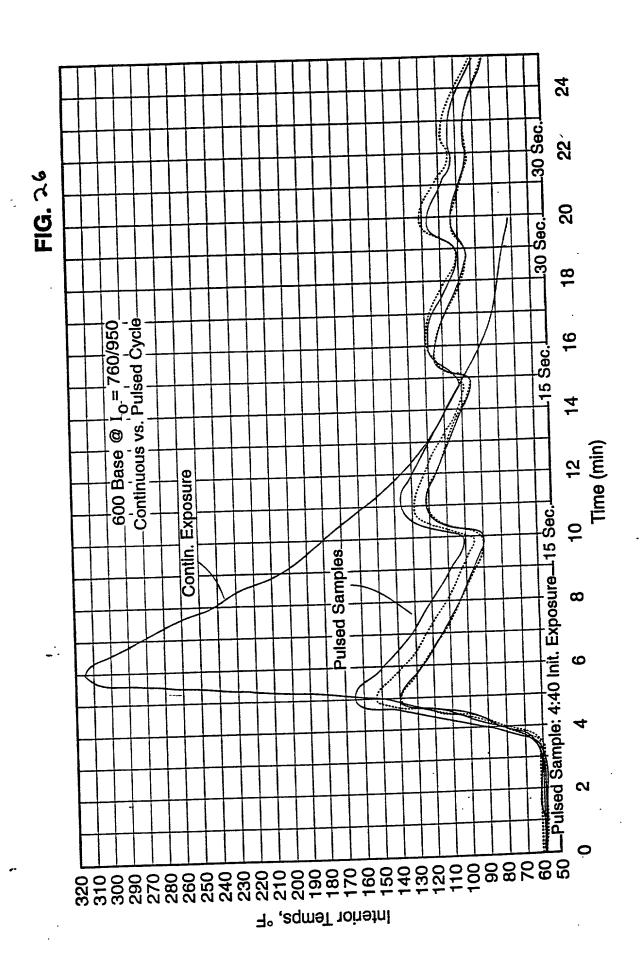
Interaction of Pulsed Method Variables (continued)

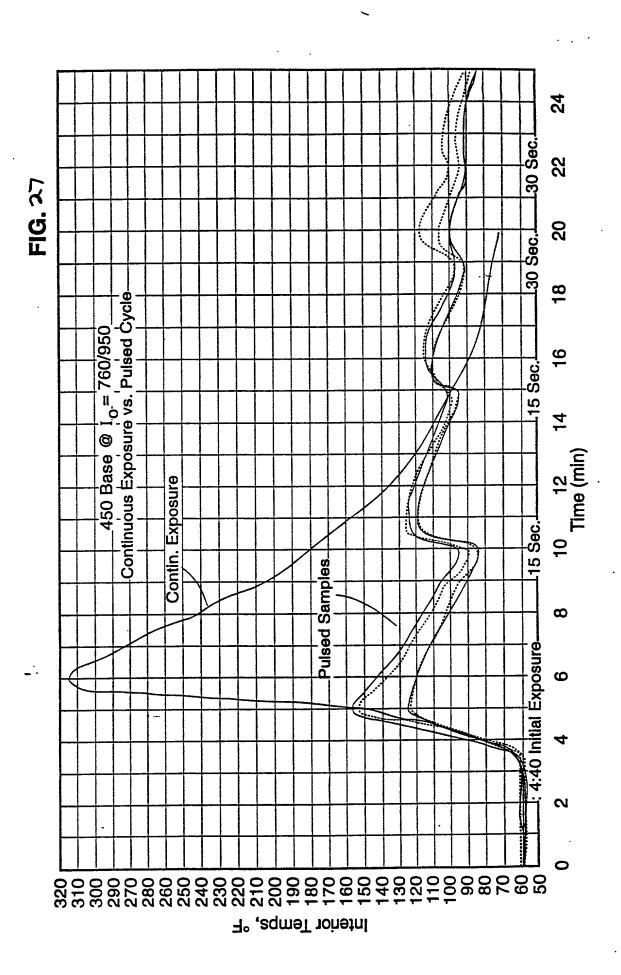
MASS OF SAMPLE	The effect that this variable will tend to have:

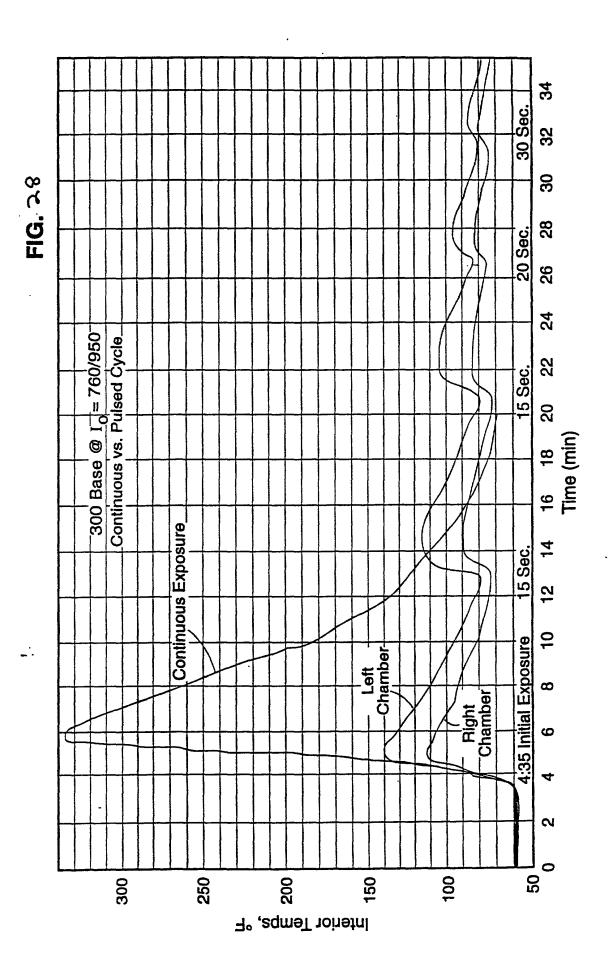
DURATION OF PULSES	The effect that to On this cycle variable in: TOTAL EXPOSURE TIME
The length of the pulses during each phase of the curing cycle may be adjusted for different mass samples. The time between pulses may be increased /decreased according to mass.	On this cycle variable in: TOTAL EXPOSURE TIME TIME Time Time Total increased sample mass tends to loresult in decreased total exposure time and a greater number of pulse/cooling cycles. Increased exposure time and to require to polymerize increased exposure time. It is believed, however, that changes in light intensities may have little impact above a certain light saturation point for the sample.
The duration of the pulses may be varied in inverse proportion with the light intensity selected. It is believed, however that changes in light intensities may have little impact above a certain the heat is light "saturation" point for the sample. A pulse will certain amo dissipated. duration ten to the time to the heat is I changes in removal shear affect the ice.	Increased light intensity will tend to result in decreased total exposure time and decreased light intensity will tend to require light intensity will tend to require increased exposure time. It is believed, however, that changes in light intensities may have little impact above a certain light "saturation" point for the sample.
A pulse will tend to generate a certain amount of heat to be dissipated. Since the pulse duration tends to be small relative to the time between pulses when the heat is being removed, changes in the rate of heat removal should not significantly affect the ideal pulse duration.	There is only a small relationship between the total dosage of light a particular mass sample requires to polymerize and the rate at which it is being cooled.
Various lens forming materials require different pulse duration depending upon their reactivity. For a selected material, slight differences in initiator & inihibitor levels will not tend to affect pulse duration.	A significant effect that monomer identify may have on total cycle time may be contributed by differences in the preferred initial exposure period. Various lens forming materials may also require longer/shorter duration pulses depending upon their reactivity.

FIG. 24 (continued)









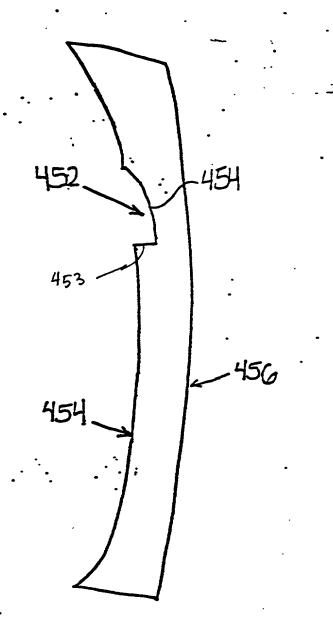
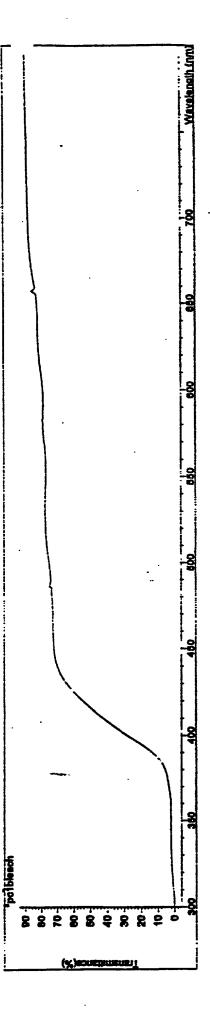
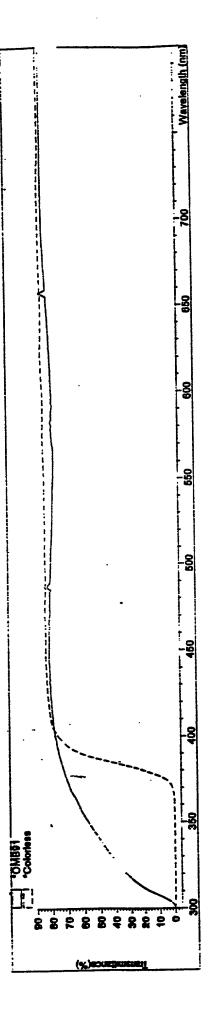


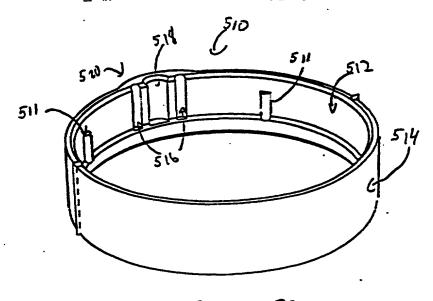
FIG. 56 29

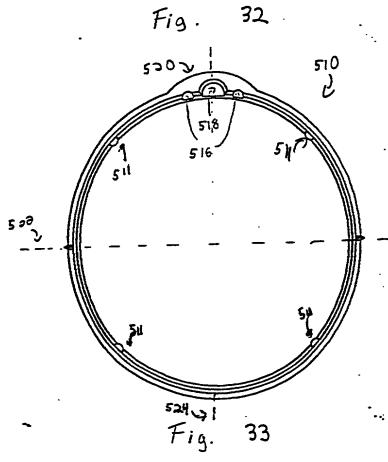


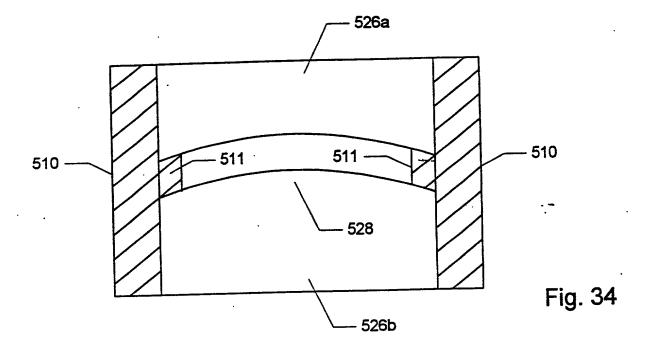
F16, 30

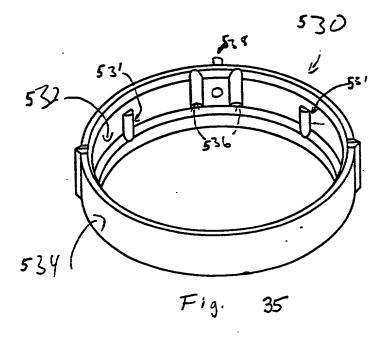


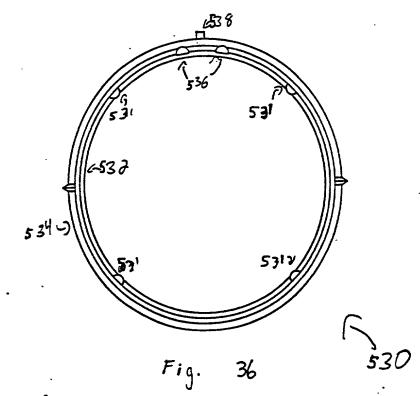
F16, 31











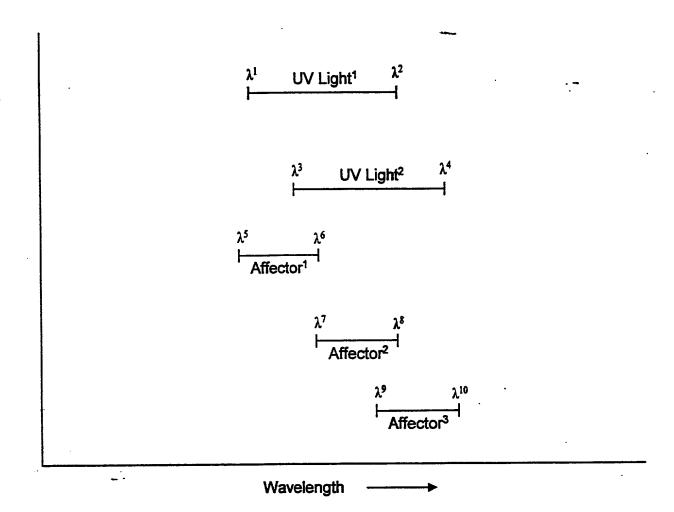
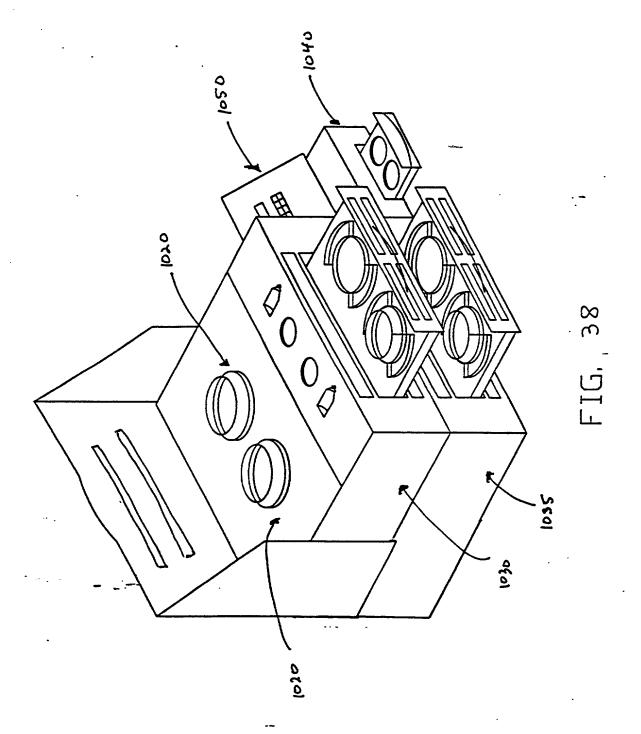


FIG. 37



$$(A) \qquad R_0 \qquad \bigcap_{n \in \mathbb{N}_2} R_1$$

$$(\beta) \qquad \stackrel{\mathsf{R}_0}{\longrightarrow} \qquad \stackrel{\mathsf{N}}{\longrightarrow} \qquad \stackrel{\mathsf{R}_1}{\longrightarrow} \qquad \stackrel{\mathsf{R}_2}{\longrightarrow} \qquad$$

$$\begin{array}{c|c} (C) & R_0 & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

F16.39

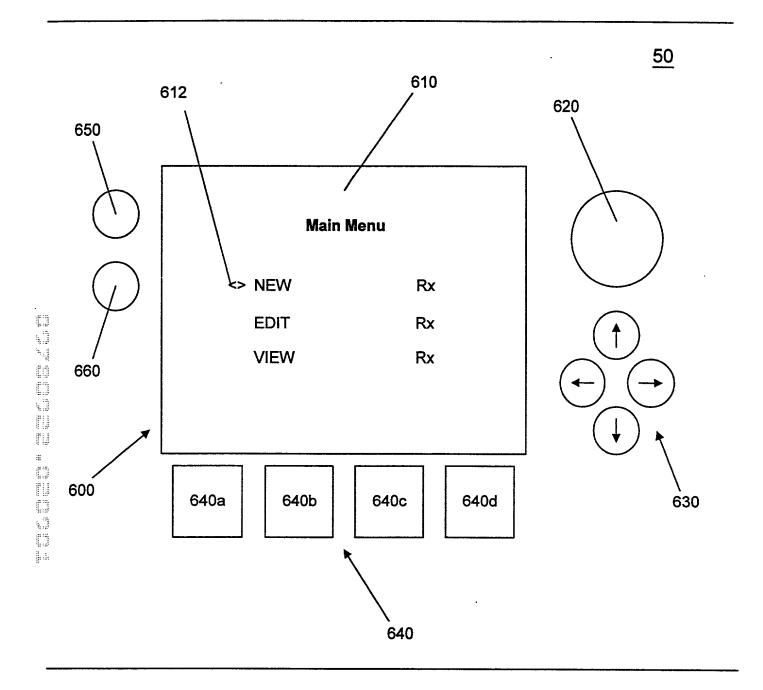


FIG. 40

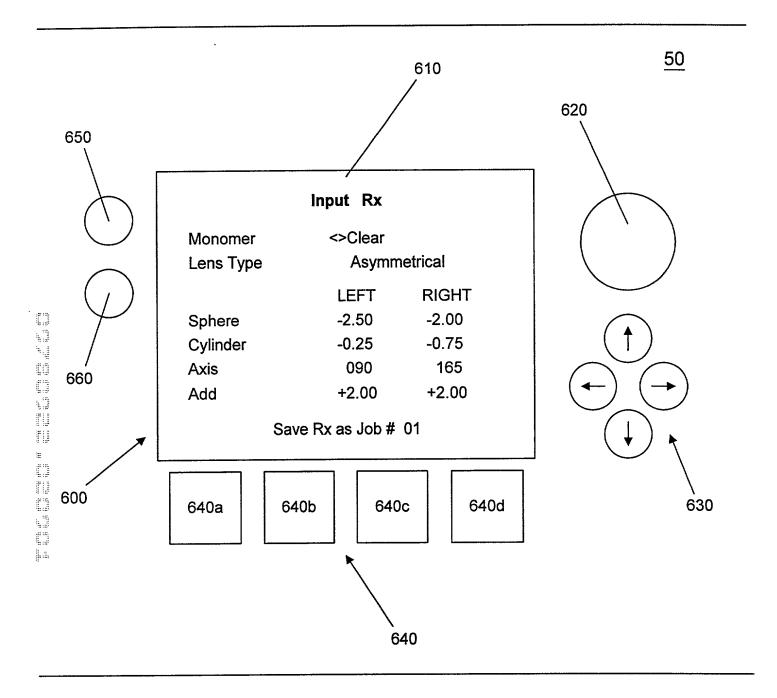


FIG. 41

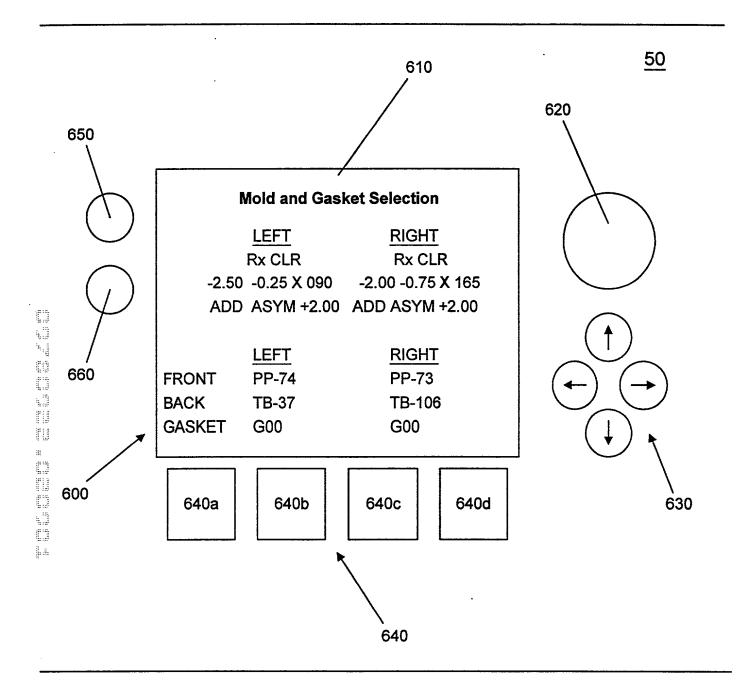


FIG. 42

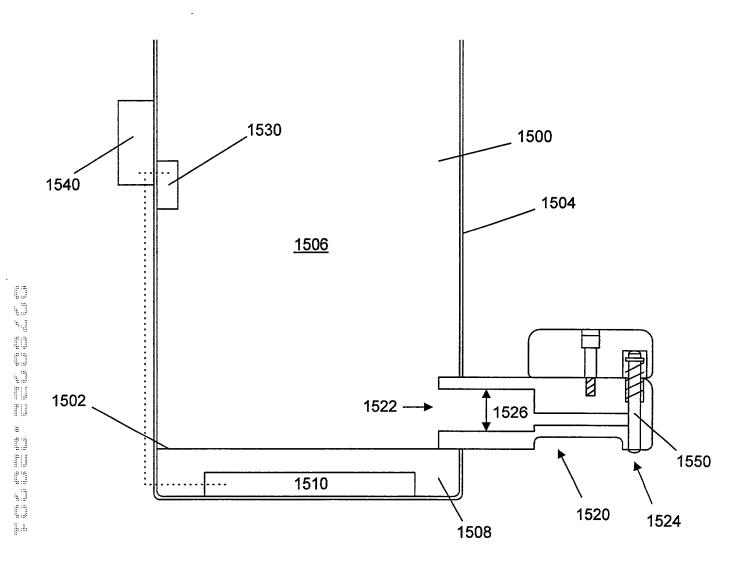


FIG. 43

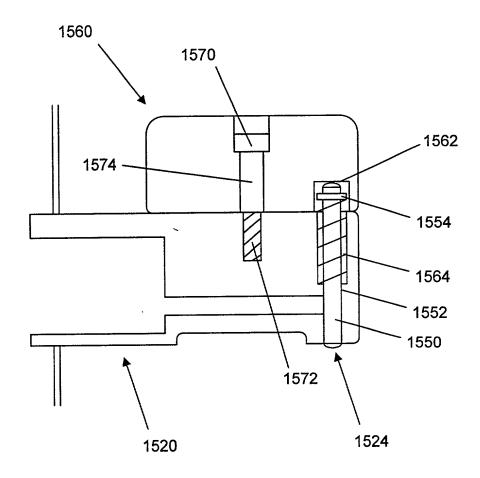


FIG. 44

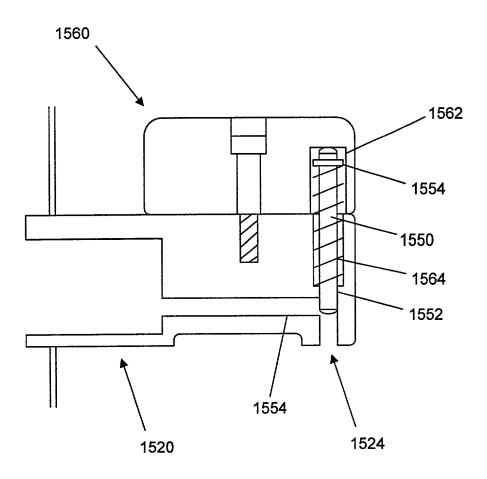


FIG. 45

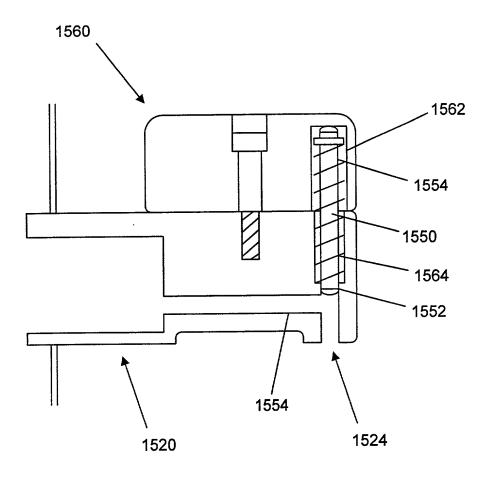
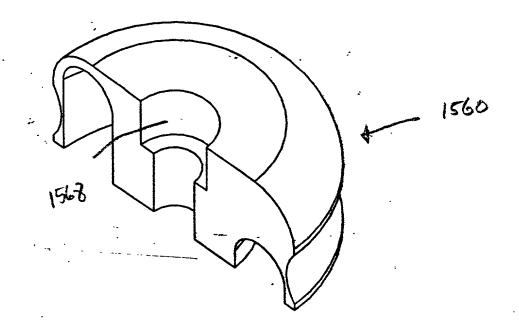
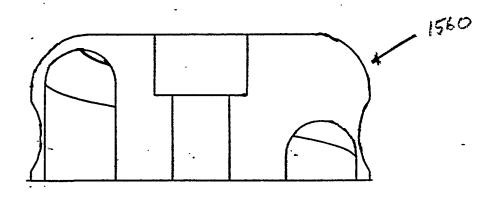


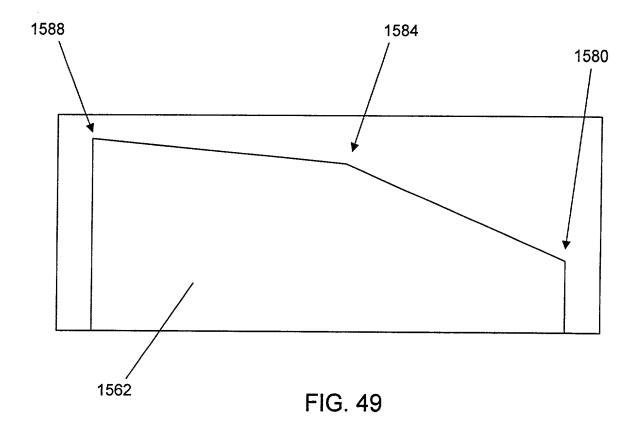
FIG. 46



F16.47



F16, 48



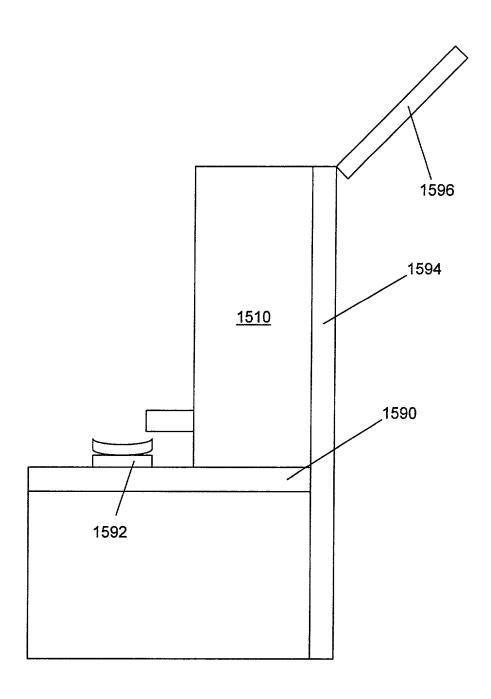


FIG. 50

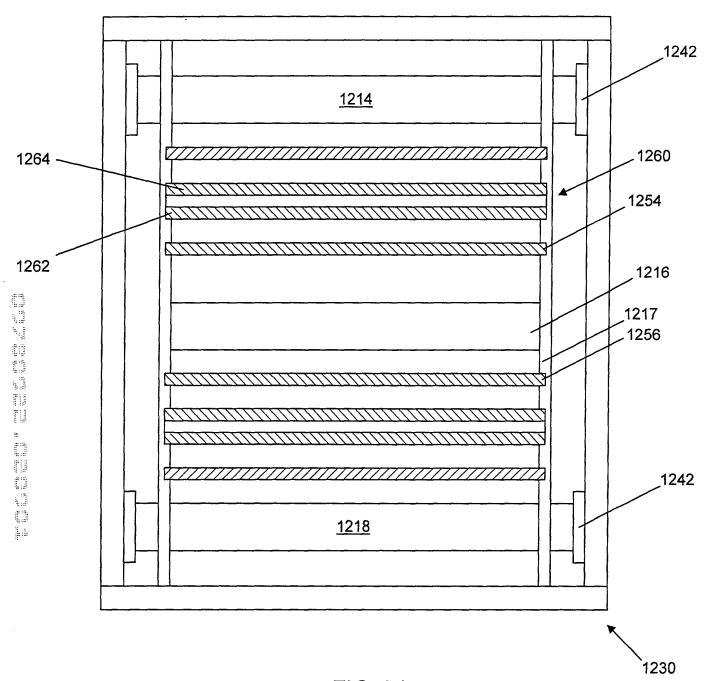


FIG. 51

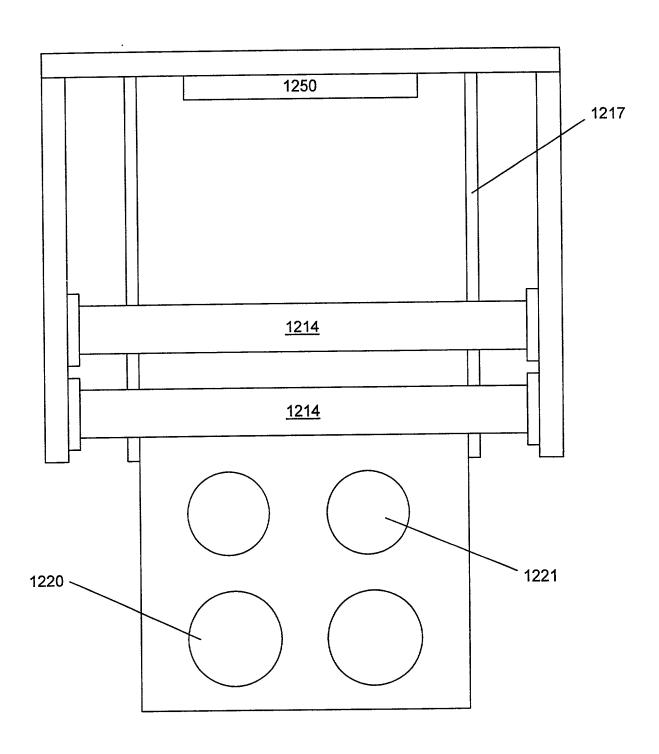


FIG. 52

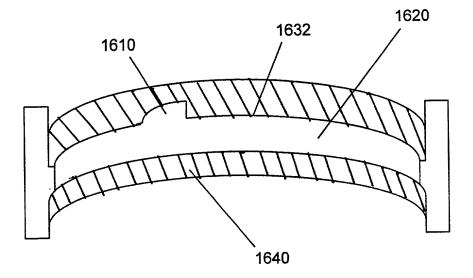


FIG. 53